

METHYL α -ACYLOXY- γ -METHYLENE- β -TETRONATE. PREPARATION AND USE AS A BUILDING BLOCK FOR THE SYNTHESIS OF THE SPIROTETRONIC ACID STRUCTURE OF CHLOROTHRICOLIDE

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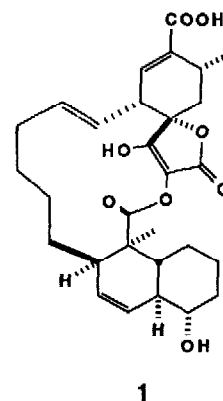
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Abstract: The Diels-Alder reaction of the title compound **4** and octatrienol **10**, which were prepared from 4-methoxy-5-methylene-2(5H)-furanone and furfuryl alcohol or D-glyceraldehyde, respectively, produced in high yield a mixture of stereo- and regio-isomeric spirotetronates: **14**, **15**, and **16** in about 65:25:10 ratio. The *exo*-mode adduct **15** (R= 2,4-difluorophenyl) was transformed into carboxylic acid **19**, the tetronic acid subunit of chlorothricolide (**1**), and also to **21** which can be utilized in the synthesis of **1**.

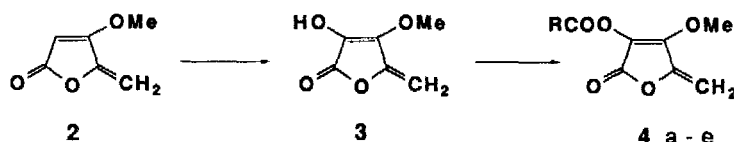
Chlorithricolide (**1**), the aglycone of the antibiotic chlorothricin reported in 1969,¹ has been the molecule of synthetic endeavors in several laboratories.^{2,3} Of the two bridging subunits in the macrolide structure, a hydronaphthalene and a spirotetronic acid, great advance has been made in the stereoselective synthesis of the former fragment (bottom-half),^{2,3} whereas synthesis of the functionalized spiro- α -hydroxytetronic acid (top-half) has remained unachieved.^{2,4}

We describe here an easy access to the top-half structure of **1** via Diels-Alder reaction of α -acyloxy- γ -methylene- β -tetronate (**4**) with an appropriate triene **10**.⁵

The dienophile (**4**) was prepared from (γ -methylene)tetronate (**2**)⁵ as shown in Scheme 1: (1) lithiation of **2** with LiN(i-Pr)₂ in THF at -90 °C followed by sequential treatment with B(OMe)₃ and H₂O₂⁶ (ca. 50% yield of **3**), and (2) O-acylation of **3** by conventional manners.⁷



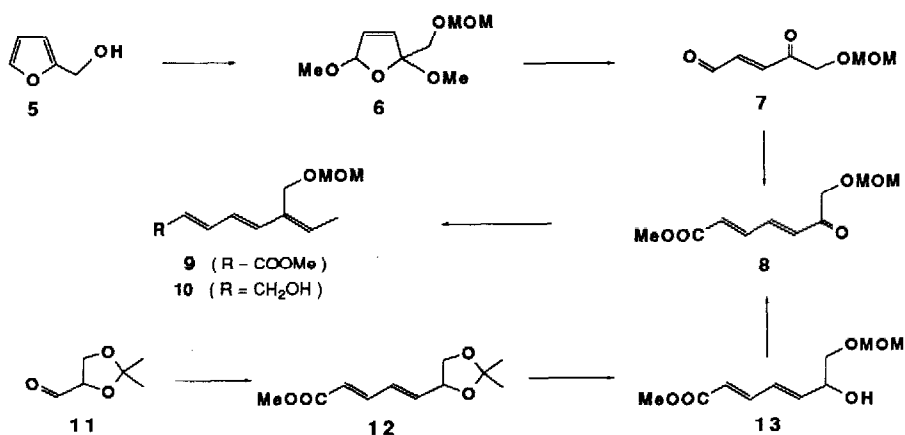
Scheme 1.



	R	mp (°C)	yield (%)
a	Me	76-7	84
b	Me ₃ C	61-3	84
c	C ₆ H ₅	136-8	79
d	2,4-F ₂ C ₆ H ₃	120-2	78
e	2,4-Cl ₂ C ₆ H ₃	118-20	83

Triene **10** was prepared by using furfuryl alcohol (**5**) as the starting material (Scheme 2). Oxidative methoxylation⁸ of **5** (Br_2 , MeOH, Et_2O) followed by \underline{O} -methoxymethylation produced dihydrofuran **6** (73% overall yield), which upon treatment with a cation exchange resin (Dowex 50x) in aqueous acetone afforded 3-oxo-2(E)-pentenal **7**. The crude aldehyde (bp 80 °C/0.2 torr) was allowed to react with $\text{Ph}_3\text{P}=\text{CHCOOMe}$ in benzene at room temperature to give keto-ester **8**, mp 40-41 °C, in 25-30% yield from dihydrofuran **6**. Wittig olefination of **8** with $\text{Ph}_3\text{P}=\text{CHMe}$ (phosphonium iodide + *n*-BuLi) in THF at -78 to 25 °C afforded trienoate **9** (85-90% diastereomeric purity) in 71% yield, which was then reduced with *i*-Bu₂AlH to give the requisite trienol **10** (92% yield). Alternatively, the intermediate **8** was prepared from D-glyceraldehyde (**11**) in 4 steps: (1) Wittig-Horner reaction with $(\text{EtO})_2\text{P}(\text{O})\text{CH}_2\text{CH}=\text{CHCOOMe}/\text{NaH}$ (57% yield of **12**), (2) deacetalization (TsOH , $\text{H}_2\text{O}-\text{MeCN}$), (3) \underline{O} -methoxymethylation (40% yield from **12**), and (4) Swern oxidation⁹ of **13** (76% yield).

Scheme 2.



The Diels-Alder reaction of **10** with **4a-e** proceeded in high yields but produced three isomeric adducts: **14**, **15**, and **16** in the ratios shown in Table 1.^{10,11} The major product in all cases was not unexpectedly⁵ the undesired stereoisomer **14** arising from an endo-mode addition, and the yield of the desired diastereomer **15** is not high (ca. 23%). Nonetheless, we believe that this synthetic approach to the top-half of **1** is attractive since the Diels-Alder reaction partners are readily accessible.

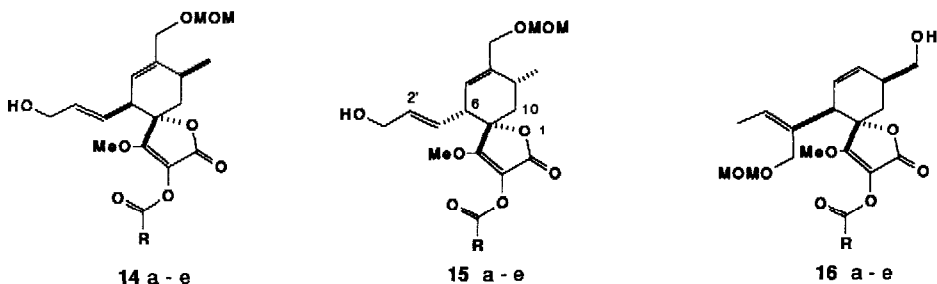


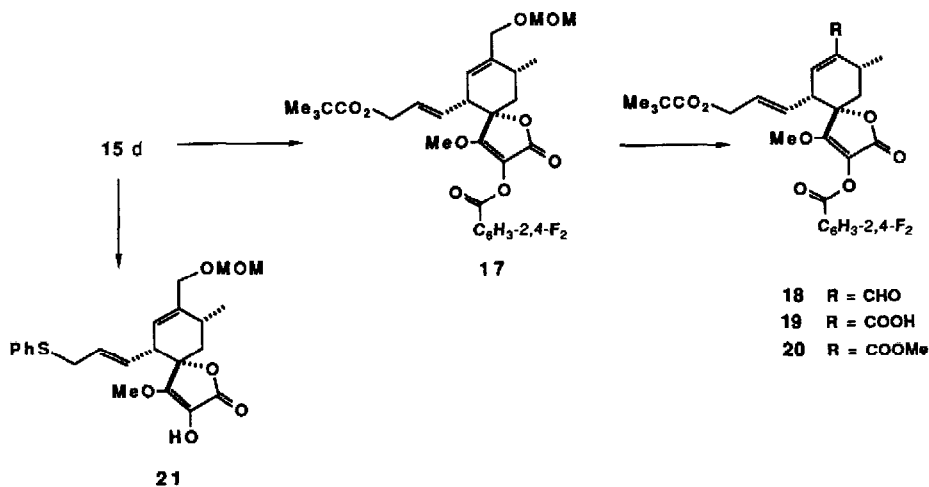
Table 1. Diels-Alder Reaction^a of **4** and **10**.^b

dienophile ^c	combined yield ^d (%)	ratio ^e of 14 , 15 , and 16	yield of 15 (%)
4a	57 (70) ^f	67 : 23 : 10	16
4b	67 (96) ^f	65 : 24 : 11	23
4c	66 (76) ^f	68 : 23 : 9	17
4d	78 (98) ^g	66 : 23 : 11	22.5
4e	79 (100) ^f	61 : 24 : 15	24

^a With chlorobenzene solvent in the presence of 4,4'-thiobis(2-*t*-butyl-6-methylphenol). ^b A ca. 6:1 mixture of **10** and its 6(E) isomer was employed. ^c 1.5 Equiv. of **4** was used. ^d Numbers in parenthesis are the yields based on recovered triene. ^e Ratios were estimated by 270 MHz ¹H-NMR spectroscopy. ^f 170-175 °C for 6 h. ^g 140-143 °C for 10 h.

With a short-step entry to the top-half structure of **1** established, we pursued conversion of the MOM-protected allyl alcohol group in **15** into carboxylic acid, the synthetic operation required in the total synthesis of **1** after construction of the macrolide nucleus. Thus, *O*-trimethylacetate (**17**) of **15d** (Scheme 3) was first subjected to removal of the MOM group by heating with LiBF₄ in wet acetonitrile¹². The resulting free alcohol (78%) was oxidized with PDC¹³ to give the corresponding aldehyde **18**, which on treatment with NaClO₂ in the presence of 2-methyl-2-butene¹⁴ afforded the carboxylic acid **19**. The structure of **19** was confirmed by leading it to the methyl ester **20** with diazomethane (48% overall yield from **17**)¹⁵. In addition, compound **15d** could be readily transformed into **21** by phenylsulfenylation of the allylic hydroxyl group (PhSSPh, *n*-Bu₃P, pyridine)¹⁶ followed by hydrolysis of the benzoate group (LiOH, MeOH). Esterification of **21** with an appropriate bottom-half carboxylic acid and subsequent macrocyclization are in progress.

Scheme 3.



References and Notes

- M. Brufani, S. Cerrini, W. Fedeli, F. Mazza, and R. Muntwyler, *Helv. Chim. Acta*, **55**, 2094(1972); R. Muntwyler and W. Keller-Schierlein, *ibid.*, **55**, 2071(1972); W. Keller-Schierlein, R. Muntwyler, W. Pache, and H. Zaehner, *ibid.*, **52**, 127(1969).
- R. E. Ireland and M. D. Varney, *J. Org. Chem.*, **51**, 635(1986).
- W. R. Roush and R. Riva, *J. Org. Chem.*, **53**, 710(1988) and references cited therein; J. A. Marshall, J. Grote, and J. E. Audia, *J. Am. Chem. Soc.*, **109**, 1186(1987); R. K. Boeckman, Jr. and T. E. Barta, *J. Org. Chem.*, **50**, 3423(1985); B. B. Snider and B. W. Burbaum, *ibid.*, **48**, 4370(1983); K. Takeda, M. Shinagawa, T. Koizumi, and E. Yoshii, *Chem. Pharm. Bull.*, **30**, 4000(1982); R. E. Ireland, W. J. Thompson, G. H. Srouji, and R. Etter, *J. Org. Chem.*, **46**, 4863(1981).
- R. R. Schmidt and R. Hirsenkorn, *Tetrahedron Lett.*, **25**, 4357(1984).
- For utilization of the α -unsubstituted compound **2** in the synthesis of the top-half of kijanolide, an analogue of chlorithricolide (**1**) having a 13-membered ketone nucleus, see K. Takeda, S. Yano, M. Sato, and E. Yoshii, *J. Org. Chem.*, **52**, 4135(1987).
- D. T. Witliak and A. K. Tehim, *J. Org. Chem.*, **52**, 2324(1987).
- The acylation was carried out in CH_2Cl_2 solvent at room temperature using the following reagents: **4a** $\text{Ac}_2\text{O}/\text{Et}_3\text{N}$; **4b** pivaloyl chloride/ Et_3N /4-dimethylaminopyridine (DMAP); **4c-e** carboxylic acid/ $\text{Et}_3\text{N}/N,N'$ -dicyclohexylcarbodiimide/DMAP.
- O. Achmatowicz, Jr., P. Bukowski, B. Szechner, Z. Zwierzchowska, and A. Zamojski, *Tetrahedron*, **27**, 1973(1971).
- K. Omura, A. K. Sharma, and D. Swern, *J. Org. Chem.*, **41**, 957(1976); S. L. Huang, K. Omura, and D. Swern, *ibid.*, **41**, 3329(1976).
- Silica gel chromatography of the product mixture provided **14**, and a mixture of **15** and **16** from which **15** was isolated after O-acylation or O-silylation. The regioisomer **16** could not be obtained in a pure state due to contamination of presumably its diastereomer.
- The stereochemistries of the Diels-Alder adducts as depicted were assigned on the basis of $^1\text{H-NMR}$ spectral analysis (270 MHz, CDCl_3). Characteristic difference between **14** and **15** is observed in the chemical shifts and coupling constants of the C(6) methine and C(10) methylene protons on the cyclohexane ring.⁵ Spectral data for the representative compounds are given below.
Compound 14d ($R=2,4$ -difluorophenyl) [R_f 0.29, hexane-AcOEt = 1:2]: $^1\text{H-NMR}$ δ 1.14 (d, $J = 7.1$ Hz, 3H, Me-9), 1.59 (br, 1H, OH), 1.82 (dd, $J = 13.9, 10.3$, 1H, H-10), 1.96 (ddd, $J = 13.9, 4.9, 1.5$ Hz, 1H, H-10), 2.67 (br m, 1H, H-9), 3.01 (br dd, $J = \text{ca. } 8$ Hz, 1H, H-6), 3.39 (s, 3H, OMe), 4.0 (d, $J = \text{ca. } 12$ Hz, 1H, CHH-8, overlapped with 4.03 singlet), 4.03 (s, 3H, OMe), 4.13 (br m, 3H, CH₂-3' and CHH-8), 4.61 and 4.67 (each d, $J = 6.6$, 1H, OCH₂OMe), 5.58 (br s, 1H, H-7), 5.59 (ddt, $J = 15.4, 8.2, \text{ca. } 1$ Hz, 1H, H-1'), 5.72 (dt, $J = 15.4, 5.0$ Hz, 1H, H-2'), 6.90-7.04 and 8.07-8.15 (m, ArH). IR (neat): 3450, 1760, 1680 cm^{-1} . Mass spectrum (EI) m/z : 480 (M^+), 419, 418, 338, 141 (base peak).
Compound 17 [R_f 0.35, hexane-AcOEt = 3:2]: $^1\text{H-NMR}$ δ 1.27 (d, $J = 7.3$ Hz, Me-9), 1.81 (dd, $J = 14.2, 1.3$ Hz, H-10), 2.32 (dd, $J = 14.2, 7.3$ Hz, H-10), 2.65 (m, H-9), 3.29 (dm, $J = \text{ca. } 7.5$ Hz, H-6), 3.39 (s, OMe), 4.01 (d, $J = 12.1$ Hz, CHH-8), 4.08 (s, OMe), 4.16 (br d, $J = 12.1$ Hz, CHH-8), 4.54 (t, $J = 5.3$ Hz, 2H, H-3'), 4.63 and 4.66 (each d, $J = 6.6$ Hz, OCH₂OMe), 5.55 (br s, H-7), 5.63 (dd, $J = 15.4, 7.5$ Hz, H-1'), 5.71 (dt, $J = 15.4, 5.3$ Hz, H-2'), 7.04-6.89 and 8.14-8.06 (m, ArH). IR (neat): 1775, 1765, 1725, 1690 cm^{-1} . Mass spectrum (EI) m/z : 564.2149 (M^+ , calcd 564.2169), 503 ($M^+ - \text{MeOCH}_2\text{O}$), 141 (base peak, difluorobenzoyl).
O-Trimethylacetate of 16d ($R=2,4$ -difluorophenyl) [R_f 0.38, hexane-AcOEt = 3:2]: $^1\text{H-NMR}$ (C_6D_6) δ 1.15 (s, t-Bu), 1.40 (dd, $J = 13.4, 5.9$ Hz, H-10), 1.62 (d, $J = 7.1$ Hz, =CHMe), 1.83 (dd, $J = 13.4, 11.4$ Hz, H-10), 2.63 (m, H-9), 3.23 (s, OMe), 3.31 (br s, H-6), 3.79 (dd, $J = 10.9, 5.8$ Hz, CHH-9), 3.83 (d, $J = 11$ Hz, OCHHC=), 3.98 (dd, $J = 10.9, 5$ Hz, CHH-9'), 4.29 (d, $J = 11$ Hz, OCHHC=), 4.53 (d, $J = 6.4$ Hz, OCHHO), 4.57 (d, $J = 6.4$ Hz, OCHHO), 5.53 (ddd, $J = 10.5, 7.4, 2.7$ Hz, H-8), 5.60 (q, $J = 7.1$ Hz, =CHMe), 5.67 (br d, $J = 10.5$ Hz, H-7), 6.2-6.3 and 7.65-7.75 (m, ArH). IR (neat): 1775, 1725, 1685 cm^{-1} . Mass spectrum (EI) m/z : 565 ($M^+ + 1$), 520, 141 (base peak).
- B. H. Lipshutz and D. F. Harrey, *Synth. Commun.*, **12**, 267(1982).
- E. J. Corey and G. Schmidt, *Tetrahedron Lett.*, 399(1979).
- G. A. Kraus and M. J. Taschner, *J. Org. Chem.*, **45**, 1175(1980).
- Spectral data of **20**: $^1\text{H-NMR}$ (270 MHz, CDCl_3) δ 1.21 (s, t-Bu), 1.31 (d, $J = 7.3$ Hz, Me-9), 1.87 (dd, $J = 14.2, 1.2$ Hz, 1H, H-10), 2.30 (dd, $J = 14.2, 7.2$ Hz, 1H, H-10), 3.00 (m, 1H, H-9), 3.36 (dt, $J = 8.2, 2.3$ Hz, H-6), 3.78 (s, COOMe), 4.09 (s, MeO-4), 4.56 (dd, $J = 13, 5.5$ Hz, 1H, H-3'), 4.62 (dd, $J = 13, 5$ Hz, 1H, H-3'), 5.64 (dd, $J = 15.5, 8.2$ Hz, H-1'), 5.77 (dt, $J = 15.5, 5$ Hz, H-2'), 6.90-7.04 and 8.06-8.14 (m, ArH). IR (neat): 1775, 1720, 1695, 1615 cm^{-1} . Mass spectrum (EI) m/e : 548 (M^+), 517, 266.
- I. Nakagawa and T. Hata, *Tetrahedron Lett.*, 1409(1975).